Materials 218/UCSB: Assignment IV

Ram Seshadri (seshadri@mrl.ucsb.edu)

Due date: February 11th 2003

- 1. Describe (using projections)
 - (a) how the ZnS (zinc blende) and the diamond structure are related.
 - (b) how the CaF₂ and fcc-Cu structures are related.
 - (c) how the CsCl and perovskite structures are related.
- 2. Os
Al has the CsCl structure with $a\sim3.1$ Å. The structure of
 OsAl_2 is as follows: SG = I4/mmm (No. 139)
 a=3.162 Å c=8.302 Å Atomx
y z
 - $\begin{array}{ccccccc} {\rm Os}^{[1]} & 0 & 0 & 0 \\ {\rm Al}^{[2]} & 0 & 0 & 0.34 \end{array}$

 $\overline{\begin{matrix} [1] \\ (0,0,0) \text{ and } (\frac{1}{2},\frac{1}{2},\frac{1}{2}) \\ [2] \\ \pm (0,0,0.34) \text{ and } \pm (\frac{1}{2},\frac{1}{2},\frac{1}{2}+0.34) \end{matrix}$

This is the structure of many disilicides (including MoSi₂).

- (a) Sketch the structure of OsAl (the CsCl structure) as projections.
- (b) Sketch the structure of $OsAl_2$ as projections
- (c) What is the Os-Al distance in OsAl
- (d) What are the Os-Al distances in $OsAl_2$ (there are two that are smaller than 3 Å)

Let the following square (on the right) represent the unit cell of OsAl:



- (e) Use such squares to sketch the structure of $OsAl_2$.¹ In your sketch, distinguish clearly between atoms sitting at y = 0 (or 1) and y = 1/2, where y is the direction perpendicular to the sheet of paper. I have done this for Al and Os in the sketch above, by using shaded circles for the former, and empty circles for the latter.
- (f) Use the following equation:

$$OsAl + OsAl_2 \longrightarrow Os_2Al_3$$

to sketch the structure of Os_2Al_3 using the same kinds of squares.

¹Hint: You should recognize that the structure of $OsAl_2$ is obtained by slicing OsAl slabs and then restacking them with a shift corresponding to body-centering.