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Submit to Justin by October 21

In-class midterm on October 21

- 1. A certain perovskite ABO_3 is cubic with a = 4.00 Å. Plot the structure in sections and use this to demonstrate that:
 - (a) The A atom has 12 O nearest neighbors. What is the A-O distance ?
 - (b) The B atom has 6 O neighbors. What is the B-O distance ?
 - (c) For a cubic perovskite, the A-O distance is $\sqrt{2}$ times the B-O distance. Verify this.
 - (d) Demonstrate using a suitable sketch, that the section in the ABO_3 structure with A atoms is the same as any section of the rock-salt (NaCl) structure.
- 2. In the following figure, label the indicated Miller planes



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- 3. Use the same coordinate system and cell as in the previous figure to sketch the following Miller planes and directions:
 - (a) Planes: $(110), (\overline{1}10), (\overline{1}11)$
 - (b) Directions: [110], [210], [123]
- 4. The compound Na_2O has a cubic unit cell. O atoms (by themselves) form an fcc structure. The Na atoms sit in the following 8 positions:

 $(\frac{1}{4}, \frac{1}{4}, \frac{1}{4}) (\frac{3}{4}, \frac{1}{4}, \frac{1}{4}) (\frac{1}{4}, \frac{3}{4}, \frac{1}{4}) (\frac{3}{4}, \frac{3}{4}, \frac{1}{4}) (\frac{1}{4}, \frac{1}{4}, \frac{3}{4}) (\frac{3}{4}, \frac{1}{4}, \frac{3}{4}) (\frac{1}{4}, \frac{3}{4}, \frac{3}{4}) (\frac{3}{4}, \frac{3}{4}, \frac{3}{4})$

Sketch the structure as sections. Do you recognize it ? What are the coordinations of Na and O ?

5. The (h, k, l) Miller planes in a tetragonal compound with a = 4.00 Å and c = 6.00 Å give rise to a series of reflections in the X-ray diffraction pattern. Using the relation (for tetragonal cells):

$$\frac{1}{d_{hkl}^2} = \frac{h^2 + k^2}{a^2} + \frac{l^2}{c^2}$$

calculate the θ values (in degrees) for which X-ray reflections will be observed corresponding to the following planes: (100), (110), (200), (222), (004). Use $\lambda = 1.5$ Å, and the Bragg law: $2d_{hkl} \sin \theta = \lambda$.