

Chemistry 2C: Assignment 1

Average: 21.6/23 (93.9%)

Common mistakes/issues:

4. Part b:

- Not calculating the numerical value of the rate (leaving it in natural log format)
 - o Points were not taken off for this
- Not explicitly noting verification of the rate relationship: $r = \ln(2)/n$
 - o Points were not taken off IF this relationship was included in their work

5. Part a:

- Not attaching all 3 plots, only attaching zero/first order, OR only attaching first order
 - o Received credit as long as plots of data were referred to in their response
- Not explicitly stating that the reaction order is identified by the plot that is linear
 - o 0.5 points removed
- Not explicitly stating that the reaction is first order
 - o 0.5 points removed

Part b:

- Including equation of linear fit BUT not specifying slope, intercept
 - o Including equation, but failing to specify slope/intercept: 0.5 points removed
 - o Writing nothing (no equation or slope/intercept): full points removed

Part c:

- Giving the rate constant, k , instead of the rate of reaction
 - o Giving k or a correct value with no units: 0.5 points removed
 - o Writing nothing: full points removed
- Failing to verify the intercept/ $[\text{H}_2\text{O}_2]_0$ relationship
 - o Most all who did this part got it correct, those who wrote nothing lost full points