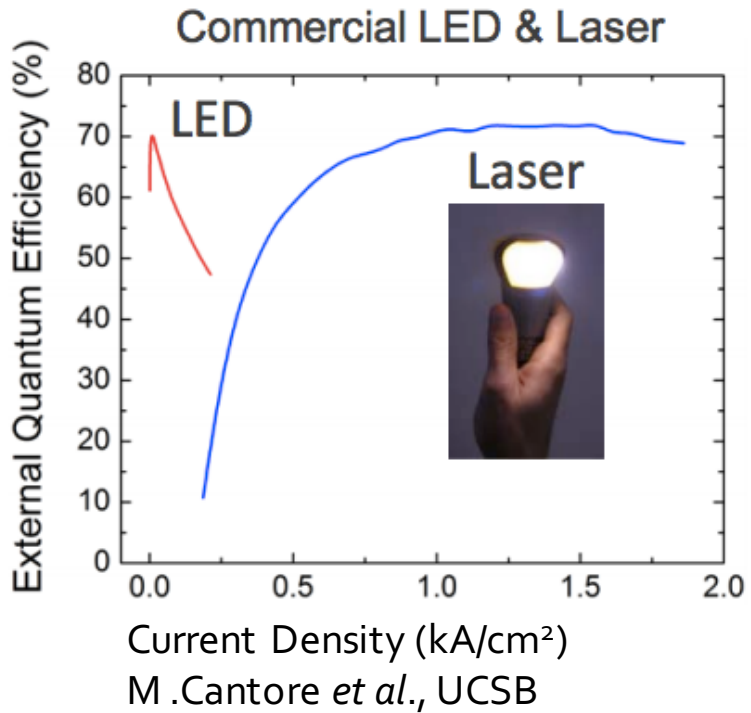


# Quantum cutting phosphors

Clayton Cozzan

# Solid state lighting is very efficient

“The laser, we believe, is the next generation of lighting, even for general applications”  
–Nakamura



**395 TWh**  
The 2030 Projected Electricity Savings from **Solid-State Lighting** equals....

**1.8X**  
2030 Projected Wind Power Electricity Generation

OR

**20X**  
2030 Projected Solar Power Electricity Generation

OR

Annual Electricity Consumption of **36 Million** U.S. Homes

OR

**\$40 Billion (US)**  
in Electricity Costs

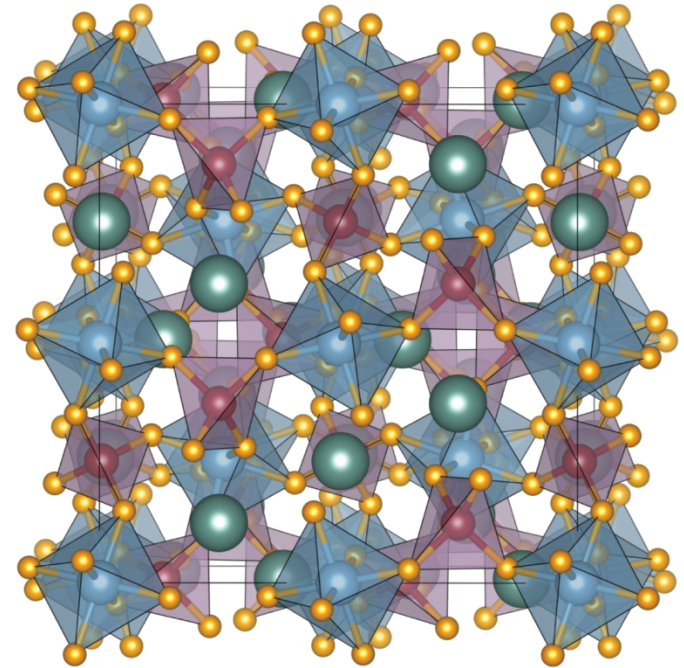
Solid state lighting research and development map, DOE, May 2015

# Phosphors are typically oxides or nitrides

Phosphors luminesce when excited by a given energy

Oxides and nitrides doped with optically active rare earth compounds

Optical properties depend on host structure and local environment



Prototypical phosphor:  
 $\text{Y}_3\text{Al}_5\text{O}_{12}$  doped with  $\text{Ce}^{3+}$

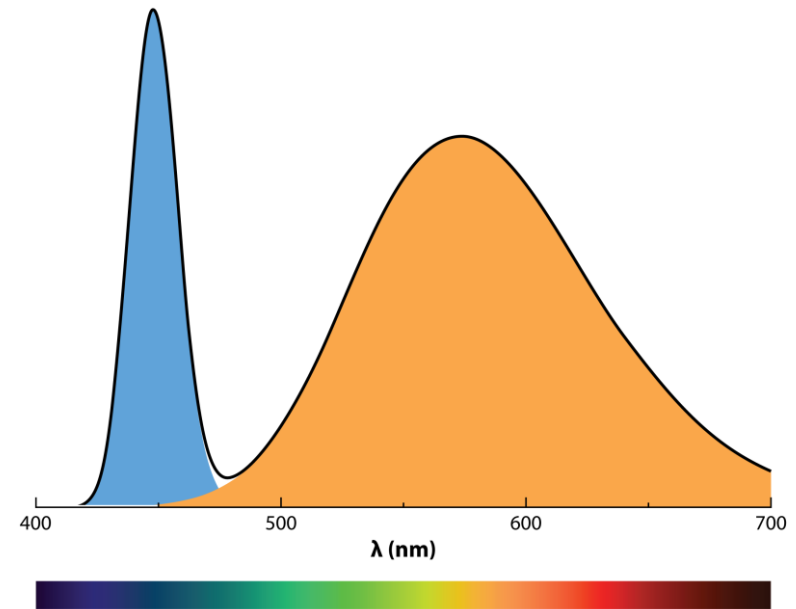
George, Denault, Seshadri, Phosphors for solid state white lighting, *Annu. Rev. Mater. Res.*, **43**, 2013

# Phosphors are wavelength converters for LEDs and laser diodes

Stokes shift: wavelength conversion loss

Typical phosphors: convert a short  $\lambda$  photon to a longer  $\lambda$  photon

Quantum cutting/splitting phosphors: convert short  $\lambda$  photon into two longer  $\lambda$  photons



George, Denault, Seshadri, "Phosphors for solid state white lighting," *Annu. Rev. Mater. Res*, **43**, 2013.

$$h\nu_1 = h\nu_2 + h\nu_3$$

energy absorbed      energy emitted

E. Fred Schubert, *Light-Emitting Diodes*, Cambridge University Press, 2006

# External quantum efficiency (EQE) describes performance

EQE is number of photons emitted by phosphor ( $s^{-1}$ ) over number of photons absorbed ( $s^{-1}$ )

$$EQE = IQE * \eta_{transfer} * \eta_{outcoupling}$$

Excitation wavelengths vacuum ultraviolet VUV (<200 nm);  
xenon discharge lamps

Normal phosphors (max): ~100%

QC phosphors (max): ~200%

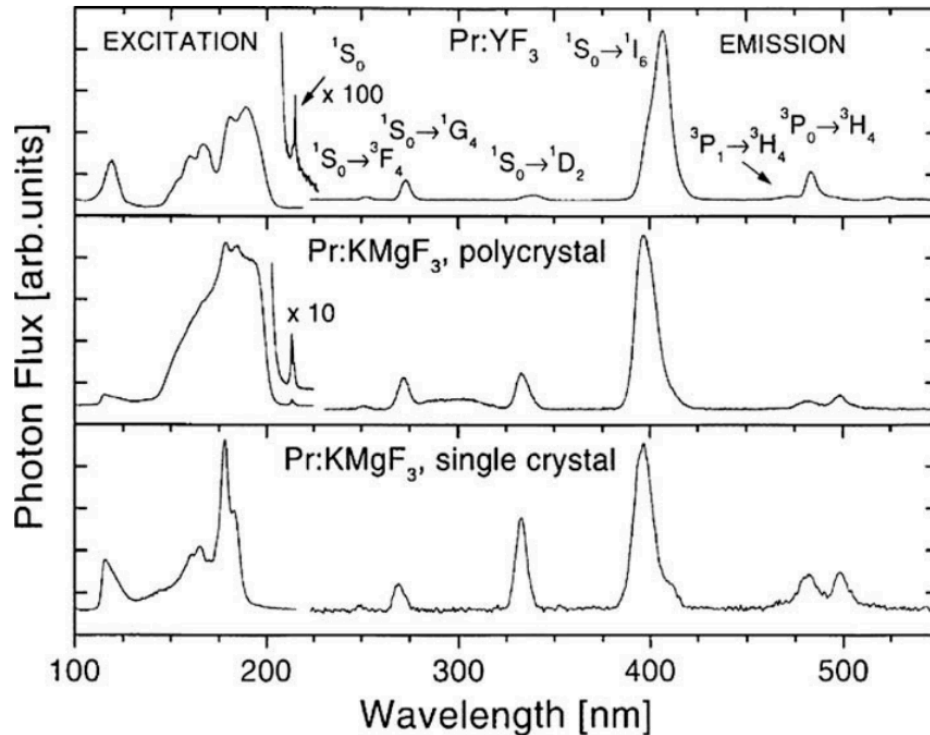
# Fluorides are used for QC phosphors

Need bandgap  $> 3.0$  eV

Excitation energy  $> 6$  eV  
( $\lambda < 200$  nm)

Energy of phonons low to  
reduce multiphonon relaxations  
(oxides are high, show no  
emission)

Downside: unstable



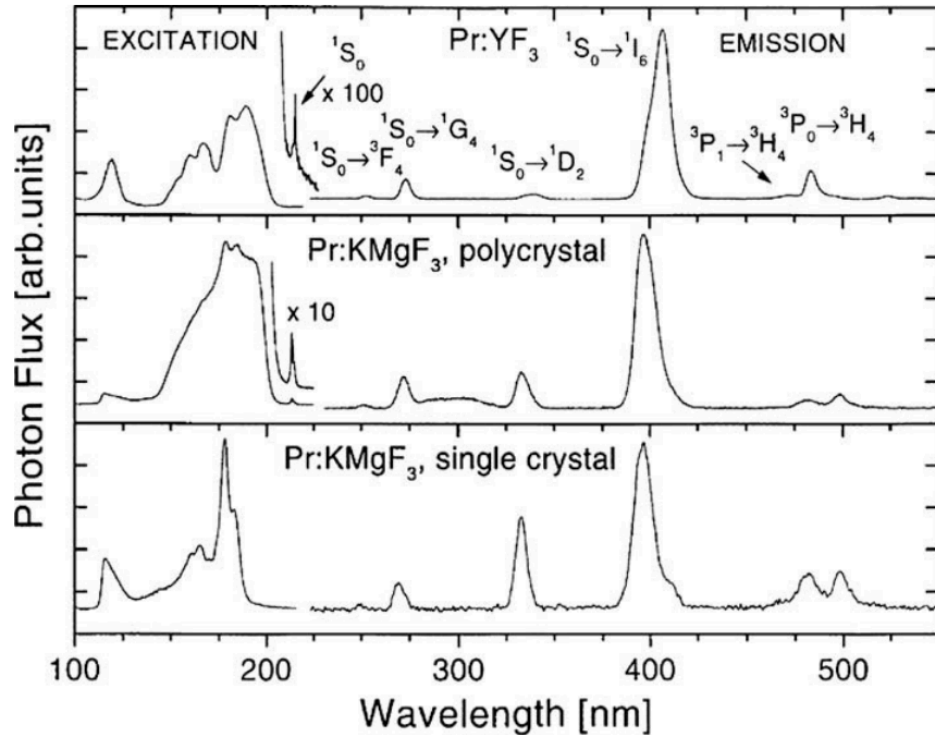
Zhang, Huang, Recent progress in quantum cutting phosphors, *Prog. Mater. Sci.*, **55** 2010

# Fluorides are used for QC phosphors

High centroid energy (occurs for compounds with high EN of anions)

Small crystal field splitting (CFS); depends on anion coordination polyhedron

Cube and octahedral coordination produce large CFS



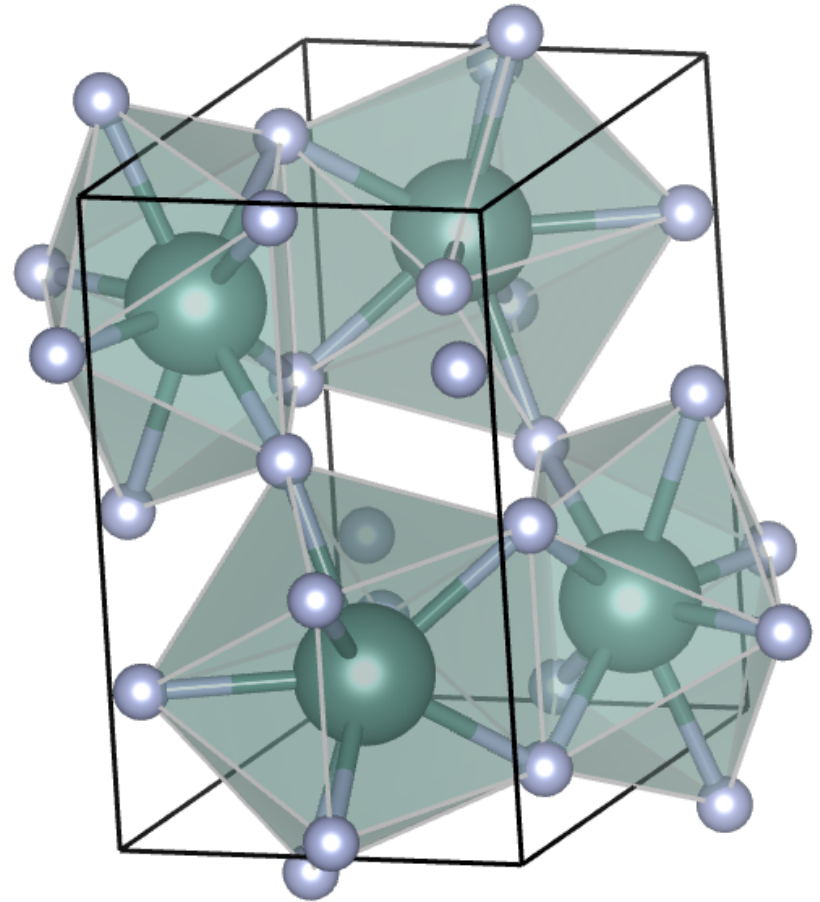
Zhang, Huang, Recent progress in quantum cutting phosphors, *Prog. Mater Sci.*, **55** 2010

# QC phosphors rely on $f$ electron transitions

Focus in on fluorides doped with  $\text{Pr}^{3+}$ ,  $\text{Tm}^{3+}$ ,  $\text{Er}^{3+}$ , and  $\text{Gd}^{3+}$

For  $\text{Gd}^{3+}$ , donor transfers stepwise to two acceptors via downconversion

$\text{Gd}^{3+}/\text{Eu}^{3+}$  pair, cascade emission

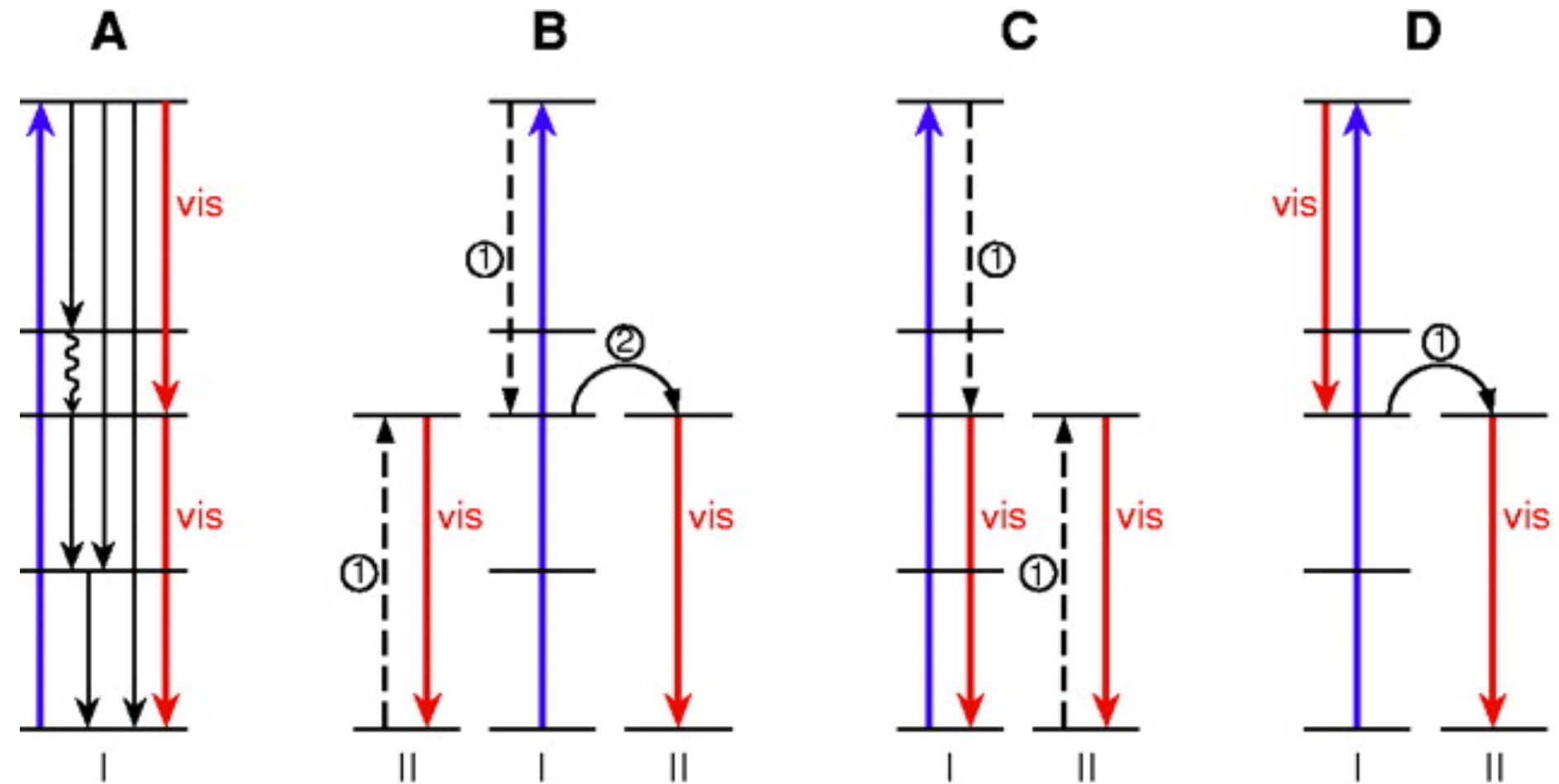


$\text{YF}_3:\text{Pr}^{3+}$ ,  $Pnma$

Structure: Cheetham, Norman, The Structures of Yttrium and Bismuth Trifluorides by Neutron Diffraction, *Acta Chem. Scand. A* **28** 1974



# QC can happen via multiple pathways



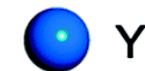
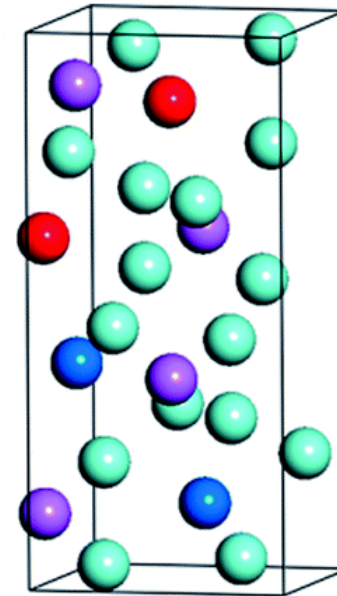
# $\text{LiGdF}_4:\text{Eu}^{3+}$ is one of the highest performing QC phosphors

Process:

Gd is excited by high-energy photon

Two step energy transfer to  $\text{Eu}^{3+}$

two visible photons emitted by  $\text{Eu}^{3+}$  (downconversion)



Wegh, Donker, Oskam, Meijerink, Visible Quantum Cutting in  $\text{LiGdF}_4:\text{Eu}^{3+}$  through downconversion, *Science* **283** 1999

Na, Jeong, Chang, Kim, Woo, Lim, Mkhoyan, Jang, Facile synthesis of intense green light emitting  $\text{LiGdY}_4:\text{Yb,Er}$ -based upconversion bipyramidal nanocrystals and their polymer composites, *Nanoscale*, **6** 2014

# Downconversion

