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# Dielectric Definition of Electronegativity

J.C. Phillips PRL 20, 550-553, 1968.

empirical

↳ Theory of covalent bonding  
PRL 19, 415-417, 1967

→  $E_h$  = covalent energy gap

C = ionic contribution  
to the energy gap

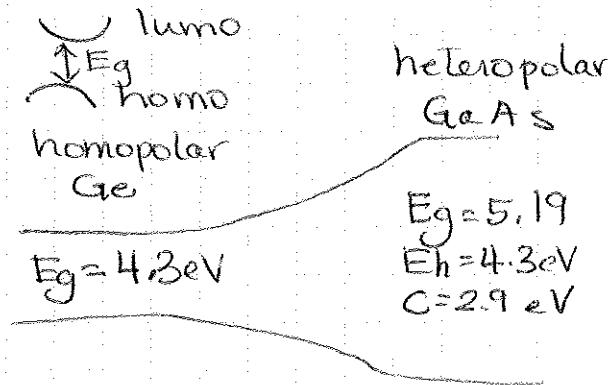
Average energy gap  $E_g$

$$E_g^2 = C^2 + E_h^2$$

$$f_i = \frac{C^2}{E_g^2} \quad \text{fraction of ionicity}$$

$$f_C = \frac{E_h^2}{E_g^2}$$

$$f_i = 0.785 \pm 0.010$$



$\epsilon_0$  = macroscopic dielectric constants

$$\text{Homo polar } \epsilon_{00} : 1 + \frac{(\hbar\omega_p)^2}{(E_g)^2} A \quad A \approx 1$$

$$\hookrightarrow \epsilon_0 = 1 + \frac{(\hbar\omega_p)^2}{[E_h^2 + C_{\alpha\beta}^2]} A \quad \hbar\omega_p = \text{plasma energy}$$

$$C_{\alpha\beta} = 1.5 \left[ \left( \frac{Z_\alpha}{r_\alpha} \right)^2 - \left( \frac{Z_\beta}{r_\beta} \right)^2 \right] \exp(-K_s R)$$

$r =$  half the bond length of the Group IV element on the same row.

$$R = \frac{r_\alpha + r_p}{2}$$

$K_s$  = Thomas-Fermi free screening wave #

$$K_s^2 = \frac{4K_f}{\pi a_0}$$

$K_f$  = fermi momentum with appropriate p

$$(K_f^3 = 3\pi^2 N)$$

Electronegativity scale:

$$X_\alpha = 3.6 \left( Z_\alpha / r_\alpha \right) (0.9) \exp(-K_s r_\alpha) + \gamma_\alpha$$

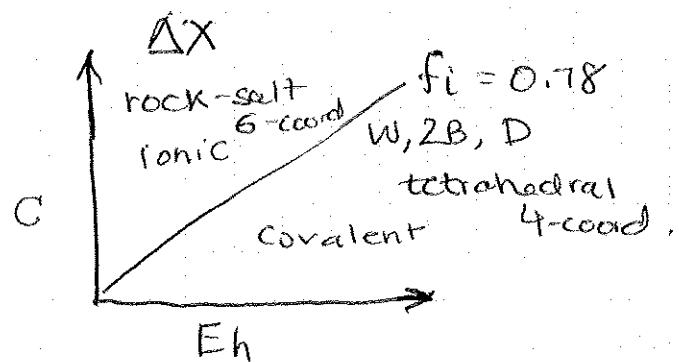
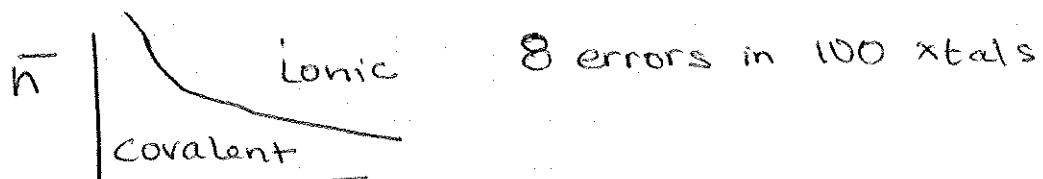
where 3.6 and  $\gamma_\alpha$  were chosen

so that  $X_\alpha(C) = 2.5$

$$X_\alpha(N) = 3.0$$

to agree with Pauling

1959 Mooser & Pearson



- much higher  $f_i$  needed for ionic transition of behavior though

i.e.  $MgO$  rock salt

refractive index decreases with hydrostatic pressure where

alkali-halides ↑

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\* What else can you predict with these?

\* Tetrahedral covalent radii  
not solely additive like Pauling's

\* Ionic radii  
— b/c different from predicted  
covalent radii

\* Nonlinear optical susceptibilities

JC Phillips JA Van Vechten  
Phys Rev (183) 709 1969

(charge transfer polarizability)

\* Melting temperatures  
JVV PRL (29) 769 1977

\* Bandgap energies  
VV Phys Rev (182) 891 1969

A 442

\* Charge distribution & piezoelectric constants  
P&VV PRL (23) 1115 1969

## References

### General Theory of Dielectric Definition of Electronegativity

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J. C. Phillips, *Chemical Physics Letters* **3**, 286 (1969).  
J. C. Phillips, J. A. Van Vechten, *Physical Review Letters* **22**, 705 (1969).  
J. C. Phillips, *Reviews of Modern Physics* **42**, 317 (1970).

### Covalent radii

- J. A. Van Vechten, J. C. Phillips, *Physical Review B* **2**, 2160 (1970).

### Melting temperature scaling

- J. A. Van Vechten, *Physical Review Letters* **29**, 769 (1972).

### Optical susceptibilities

- J. C. Phillips, J. A. Van Vechten, *Physical Review* **183**, 709 (1969).

### Piezoelectric constants

- J. C. Phillips, J. A. Van Vechten, *Physical Review Letters* **23**, 1115 (1969).